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SULFUR AS SUPPORTING ELEMENT IN VINYLCYCLOPROPANE FORMATION

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 $\frac{\text{Abstract}}{\text{obtained}} \quad \text{Vinylcyclopropanes 5,9,10 are conveniently } \\ \frac{\text{obtained}}{\text{obtained}} \quad \text{by sulfur-mediated S}_{N} \text{i reactions.}$

Bishomoallyl alcohols 3 are readily accessible from epoxides 1 and the sulfur-stabilized anion 2 of allyl phenyl sulfide. Tosylated derivatives 4 are available by tosylation of 3 or by quenching the reaction mixture from 1 and 2 with tosyl chloride. Now again the anion-stabilizing effect of sulfur can be used in a sequence of deprotonation/intramolecular nucleophilic attack giving vinylcyclopropanes 5.1

Vinylcyclopropane formation is also possible by intramolecular ring-opening of epoxides. Here, oxiranes 6 with w-tosyloxy-substituted side-chain are employed.
Reaction with anion 2 regenerates an epoxide 8 via intermediate 7. Subsequent ring-closure to vinylcyclopropanes 9,10 is then possible by sulfur-supported deprotonation or by reductive desulfurization employing the radical anion of di-tert.-butylbiphenyl (DBB).²

The methodology could be used in an asymmetric synthesis of the monoterpene ether artemeseol (11) from vinylcyclopropane 12 with oxirane 13 being the key intermediate. Also the responding nor-derivative can be obtained.³

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